

OXIDATIVE DECARBOXYLATION; FACILE ROUTE TO 18-NOR STEROIDS

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Previous methods for the preparation of 18-nor-steroids have either involved total synthesis,¹⁻³ cleavage and regeneration of the D ring,^{4,5} transformation of natural product⁶ or elaboration of the 18,20-hemiketal moiety.^{7,8} We wish to report here an alternate shorter and more efficient route, incorporating decarboxylation of the keto acid (3) as the key reaction.

Treatment of 3 β -methoxy-5-pregnen-20 β -ol⁹ with Pb (OAc)₄/HNO₃ followed by chromic acid oxidation of the crude product,¹⁰ gave the lactone (2) in 42% yield: ir (cm⁻¹) 1755; Anal (C₂₂H₃₂O₃) C,H. Alternatively, hydrolysis (K₂CO₃, MeOH, room temperature, 4h) and methylation (MeI, NaH, DME, 60 $^{\circ}$, 4h) of the readily available acetoxy lactone (1)¹⁰ afforded (2) in 75% yield. Vigorous hydrolysis (30% KOH, MeOH, reflux, 4h) of (2) and subsequent oxidation,¹¹ yielded 83% of the keto acid (3) which exists mainly as the corresponding lactol (4)¹²: ir (cm⁻¹) 3400, 1780; Anal (C₂₂H₃₂O₄) C,H.

Decarboxylation of (3) with lead tetraacetate in the presence of cupric acetate,¹³ afforded the 18-nor-steroids (5) and (6) in yields very dependent on the solvent used. When the reaction was conducted in benzene, (5) was isolated in 57% yield: ir (cm⁻¹) 1700; NMR (CDCl₃, δ) 1.0 (s, 3H-CH₃), 2.1 (s, 3H, CH₃-C=O), 3.4 (s, 3H, CH₃-O), 5.4 (m, 1H, C=CH-); Anal (C₂₁H₃₀O₂) C,H, and (6) in 11% yield. Use of HMPA as solvent however, gave (6) 50-58% ir (cm⁻¹) 1675, 1655, 1615; NMR (CDCl₃, δ) 1.0 (s, 3H, CH₃-), 2.2 (s, 3H, CH₃-C=O), 3.4 (s, 3H, CH₃-O), 5.4 (1H, M, C=CH-); uv (EtOH, λ) 257, ϵ 10,900; high resolution mass spectrum M⁺314.2247 (Calcd for C₂₁H₃₀O₂ 314.2246) and 5-10% yield of (5). Other solvents evaluated included DMF, DMSO and DME. Attempted isomerization of (5) \rightarrow (6) under acidic or basic conditions proved unsuccessful in accord with previous investigations in the bicyclo [4.3.0] nonane system.¹⁴

The 18-nor-steroid (6) is a potentially important precursor for future isotopic labelling experiments at C-18 via 1,4-addition routes.

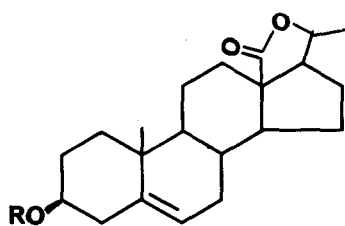
Procedure for (3) \rightarrow (6). Lead tetraacetate (200 mg, 0.45 mmol) and cupric acetate (10 mg, 0.05 mmol) were added to a degassed solution of (3) (100 mg, 0.28 mmol) in HMPA (1 ml) and pyridine (0.1 ml), followed by stirring at 80 $^{\circ}$ for 4 h. The mixture was then diluted with heptane : ethyl acetate (1.5 ml, 3:1 v/v) and filtered through Sephadex LH20 (4g), preswollen in the same solvent system. Further purification by preparative TLC (silica, light petroleum : ether 3:2 v/v) yielded (6) (50 mg, 57%) as an oil, which slowly solidified on standing at 0 $^{\circ}$. The white solid m.p. 78-80 $^{\circ}$ was >97% pure (GC).

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References and Notes

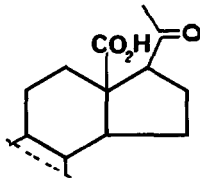
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12. This equilibrium is readily reversible e.g. treatment of (4) with diazomethane smoothly affords the respective methyl ester of (3).
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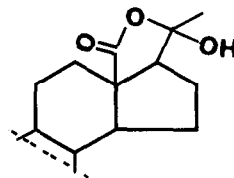


(1) R = CH₃CO

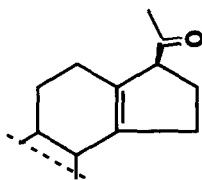
(2) R = CH₃



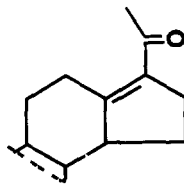
(3)



(4)



(5)



(6)